

## 8-8 Test Method – Formed Foil for Cathode 1 of 2

### 1. Scope of Test

$0.9 \leq V_{fs} \leq 5.0$

### 2. Definitions of Technical Codes

- 1)  $V_{fs}$  : Nominal Formation Voltage
- 2)  $V_t$  : Dielectric Withstanding Voltage

### 3. Test Procedure

- 1) Procedure ①Capacitance Measurement  
② $V_t$  Measurement

### 4. Test Equipment for Capacitance Measurement

- 1) Capacitance Measurement Device  
Capacitance Meter in accordance with JIS C 5101-1,4.7  
Measurement Frequency :  $120\text{Hz} \pm 5\%$   
Measurement Voltage :  $0.5V_{rms}$  or less
- 2) Measurement Vessel  
Material : Glass  
Volume : 200ml or 300ml
- 3) Counter Electrode  
Test Specimens

### 5. Test Equipment for $V_t$ Measurement

- 1) DC Power Supply  
Ripple Content : 2% or less for 50,60Hz  
1% or less for 100,120Hz  
DC Voltage Stability :  $\pm 3\%$
- 2) DC Voltmeter  
Internal Resistance :  $1M\Omega$  or higher  
Accuracy :  $\pm 0.5\%$
- 3) DC Ammeter  
Internal Resistance shall be sufficiently small compared to Load Resistance ( $10\Omega$  or less)
- 4) Measurement Vessel  
Material : SUS304  
Volume :  $500 \pm 50\text{ml}$   
Depth :  $100 \pm 20\text{mm}$
- 5) Counter Electrode  
Measurement Vessel : SUS304
- 6) Referential Electrode  
Material : Platinum Plate  
Purity : 99.99% or more  
Dimensions :  $10 \times 20 \times 0.1\text{mm}$
- 7) Volt Recorder  
Internal Resistance :  $1M\Omega$  or higher  
Accuracy :  $\pm 0.5\%$

### 6. Test Specimen

Test Specimen : In accordance with 8-1 *Selecting Test Specimens for Electrical Characteristics Measurements.*

### 7. Capacitance Measurement

- 1) Electrolyte for Capacitance Measurement  
Ammonium Adipate : 150g  
Deionized Water : 1,000ml  
Specific Resistance :  $6.5(+2.0 - 1.5)\Omega \text{ cm}/70 \pm 2^\circ\text{C}$   
pH :  $6.7(+0.5 - 1.5) / 50 \pm 2^\circ\text{C}$

#### 2) Condition for Capacitance Measurement

Measurement Temperature :  $30 \pm 2^\circ\text{C}$

#### 3) Measurement Circuit

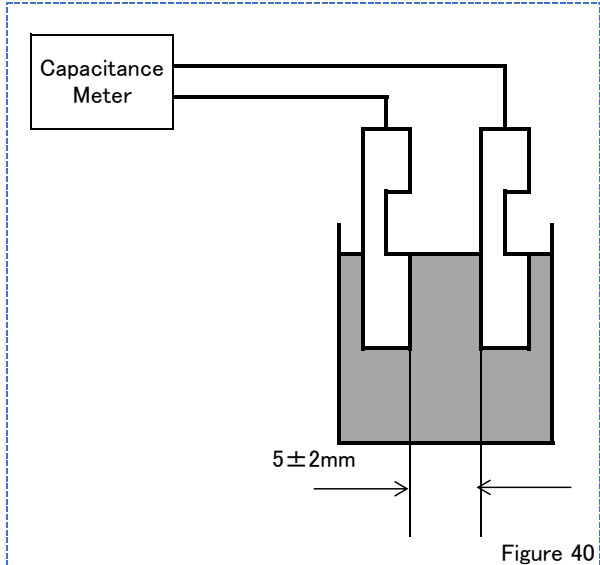


Figure 40

The test specimen shall be immersed in the measuring electrolyte so that the top edge of the Projected Area (area to be measured) is level with the surface.

#### 4) Measurement Calculation

Capacitance per  $1\text{cm}^2$  is calculated by the following formula

$$C = \frac{C_m \times 2}{5}$$

Where :  $C_m$  = Measured Value ( $\mu\text{F}$ )

:  $C$  = Capacitance ( $\mu\text{F}/\text{cm}^2$ ) per  $1\text{cm}^2$

## 8. Vt Measurement

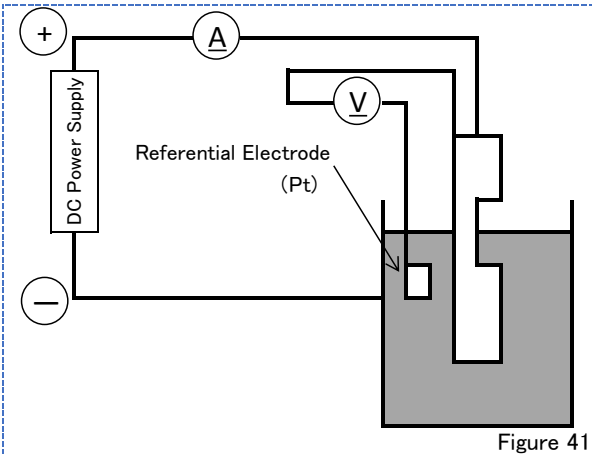
### 1) Electrolyte for Vt Measurement

Ammonium Adipate	: 150g
Deionized Water	: 1,000ml
Specific Resistance	: $6.5(+2.0 -1.5) \Omega \text{ cm} / 70 \pm 2^\circ \text{C}$
pH	: $6.7(+0.5 -1.5) / 50 \pm 2^\circ \text{C}$

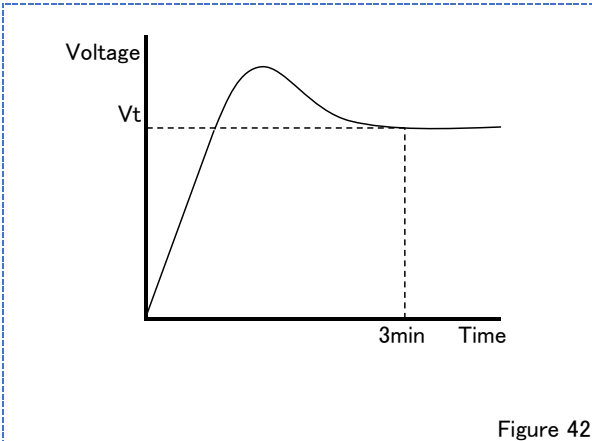
### 2) Conditions for Vt Measurement

Measurement Temperature	: $85 \pm 2^\circ \text{C}$
Current Density	: $0.5 \pm 0.05 \text{ mA} / 1 \text{ Test Specimen } 5 \text{ cm}^2$

### 3) Measurement Circuit



### 4) Measurement Method



- ① The test specimen shall be immersed in the measuring electrolyte so that the top edge of the Projected Area (area to be measured) is 6-8mm below the surface.
- ② The increase in voltage shall be measured after applying constant DC current.
- ③ Dielectric Withstanding Voltage (Vt) shall be defined as the voltage measured 3 minutes after applying DC current.

## 9. Determination

- 1)  $0.9 V_{fs} : \geq -1.1 \text{ V}$  acceptable
- 2)  $1.2 V_{fs} : \geq -0.7 \text{ V}$  acceptable